

*Annual*  
**WATER**  
**QUALITY**  
**REPORT**  
*Reporting Year 2013*



*Presented By*



PWS ID#: 1510019

Este informe contiene información muy importante sobre su agua potable. Tradúzcalo o hable con alguien que lo entienda bien.

## There When You Need Us

We are once again proud to present our annual water quality report covering all testing performed through December 31, 2013. Over the years, we have dedicated ourselves to producing drinking water that meets all state and federal standards. We continually strive to adopt new methods for delivering the best-quality drinking water to you. As new challenges to drinking water safety emerge, we remain vigilant in meeting the goals of source water protection, water conservation, and community education while continuing to serve the needs of all our water users.

Please remember that we are always available to assist you should you ever have any questions or concerns about your water.

## Source Water Assessment

An assessment of the drinking water sources for the City of Shafter was initially completed by the State of California Department of Health Services in September 1999 and updated by the City in 2009. A copy of the complete assessment is available at the City of Public Works Department, located at 336 Pacific Avenue. You may request a summary of the assessment by contacting the department at (661) 746-5002.

## Important Health Information

Some people may be more vulnerable to contaminants in drinking water than the general population. Immunocompromised persons such as persons with cancer undergoing chemotherapy, persons who have undergone organ transplants, people with HIV/AIDS or other immune system disorders, some elderly, and infants may be particularly at risk from infections. These people should seek advice about drinking water from their health care providers. The U.S. EPA/CDC (Centers for Disease Control and Prevention) guidelines on appropriate means to lessen the risk of infection by *Cryptosporidium* and other microbial contaminants are available from the Safe Drinking Water Hotline at (800) 426-4791 or <http://water.epa.gov/drink/hotline>.

## Substances That Could Be in Water

The sources of drinking water (both tap water and bottled water) include rivers, lakes, streams, ponds, reservoirs, springs, and wells. As water travels over the surface of the land or through the ground, it dissolves naturally occurring minerals and, in some cases, radioactive material, and can pick up substances resulting from the presence of animals or from human activity.

In order to ensure that tap water is safe to drink, the U.S. Environmental Protection Agency (U.S. EPA) and the California Department of Public Health (Department) prescribe regulations that limit the amount of certain contaminants in water provided by public water systems. Department regulations also establish limits for contaminants in bottled water that must provide the same protection for public health. Drinking water, including bottled water, may reasonably be expected to contain at least small amounts of some contaminants. The presence of contaminants does not necessarily indicate that water poses a health risk.

Contaminants that may be present in source water include:

**Microbial Contaminants**, such as viruses and bacteria, that may come from sewage treatment plants, septic systems, agricultural livestock operations, and wildlife;

**Inorganic Contaminants**, such as salts and metals, that can be naturally occurring or can result from urban stormwater runoff, industrial or domestic wastewater discharges, oil and gas production, mining, or farming;

**Pesticides and Herbicides**, that may come from a variety of sources such as agriculture, urban stormwater runoff, and residential uses;

**Organic Chemical Contaminants**, including synthetic and volatile organic chemicals, which are by-products of industrial processes and petroleum production, and which can also come from gas stations, urban stormwater runoff, agricultural applications, and septic systems;

**Radioactive Contaminants**, that can be naturally occurring or can be the result of oil and gas production and mining activities.

More information about contaminants and potential health effects can be obtained by calling the U.S. EPA's Safe Drinking Water Hotline at (800) 426-4791.

## Lead in Home Plumbing

If present, elevated levels of lead can cause serious health problems, especially for pregnant women and young children. Lead in drinking water is primarily from materials and components associated with service lines and home plumbing. We are responsible for providing high-quality drinking water, but cannot control the variety of materials used in plumbing components. When your water has been sitting for several hours, you can minimize the potential for lead exposure by flushing your tap for 30 seconds to 2 minutes before using water for drinking or cooking. If you are concerned about lead in your water, you may wish to have your water tested. Information on lead in drinking water, testing methods, and steps you can take to minimize exposure is available from the Safe Drinking Water Hotline or at [www.epa.gov/safewater/lead](http://www.epa.gov/safewater/lead).

## Community Participation

The City Council of the City of Shafter meet on the first and third Tuesdays of each month at 7:00 pm in the council chambers located at City Hall, 336 Pacific Avenue, to discuss and take action on various matters that affect the community. At times, issues regarding water system improvements or quality are on the agenda. Public input is appreciated and is taken into consideration during the discussions and formal actions.

## Water Conservation

You can play a role in conserving water and saving yourself money in the process by becoming conscious of the amount of water your household is using and by looking for ways to use less whenever you can. It is not hard to conserve water. Here are a few tips:

- Automatic dishwashers use 15 gallons for every cycle, regardless of how many dishes are loaded. So get a run for your money and load it to capacity.
- Turn off the tap when brushing your teeth.
- Check every faucet in your home for leaks. Just a slow drip can waste 15 to 20 gallons a day. Fix it and you can save almost 6,000 gallons per year.
- Check your toilets for leaks by putting a few drops of food coloring in the tank. Watch for a few minutes to see if the color shows up in the bowl. It is not uncommon to lose up to 100 gallons a day from an invisible toilet leak. Fix it and you save more than 30,000 gallons a year.
- Use your water meter to detect hidden leaks. Simply turn off all taps and water using appliances. Then check the meter after 15 minutes. If it moved, you have a leak.

## Where Does My Water Come From?

The source of Shafter's drinking water is an underground aquifer that is pumped to the surface by a system of ground water wells. The aquifer is replenished through the natural runoff from the Sierra Nevada Mountains as well as through seepage from the many irrigation canals that import water into the area from other regions of the state.

The City of Shafter owns and operates your domestic water supply and distribution systems. These systems operate as one of the enterprises under the City's umbrella. The water system within the core City comprises seven active ground water wells, four above-ground water storage tanks with booster pumps, and approximately 70 miles of water distribution lines. Distribution is defined as the combination of tanks, water mains, and booster pumps necessary to deliver water to our customers.

## QUESTIONS?

For more information about this report, or for any questions relating to your drinking water, please call Public Works Director Michael James at (661) 746-5002, or write to the Department at 336 Pacific Avenue, Shafter, CA 93263.

## Water Treatment Process

We are aware that some groundwater contamination does exist in the Shafter area. The ground water quality issues of primary concern to us here in Shafter include salt intrusion primarily from agricultural activities, nitrate contamination from natural and agricultural activities, organic chemical contamination also from agricultural pesticide and fumigants, and arsenic.

The only water treatment currently required of the City's water supply is disinfection by chlorination for microbiological contaminants. We test our water regularly for coliform bacteria and it is detected occasionally, but its detection is normally remedied by adjusting the chlorine dosage.

In 2009, we started using our first carbon treatment plant at one of our wells to help us learn how to remove an organic contaminant known as 1,2,3-trichloropropane. Commonly referred to as "TCP," this contaminant is not currently regulated by the state but we are aware that it will be as early as this year. In preparation for this new regulation, we are currently working to construct carbon plants at all of our active wells with the intent of fully removing TCP from detectable levels. More information on TCP detections and health risks can be found at the following state web site link: <http://www.cdph.ca.gov/certlic/drinkingwater/Pages/123TCP.aspx>.

## Benefits of Chlorination

Disinfection, a chemical process used to control disease-causing microorganisms by killing or inactivating them, is unquestionably the most important step in drinking water treatment. By far, the most common method of disinfection in North America is chlorination.

Before communities began routinely treating drinking water with chlorine (starting with Chicago and Jersey City in 1908), cholera, typhoid fever, dysentery, and hepatitis A killed thousands of U.S. residents annually. Drinking water chlorination and filtration have helped to virtually eliminate these diseases in the U.S. Significant strides in public health are directly linked to the adoption of drinking water chlorination. In fact, the filtration of drinking water plus the use of chlorine is probably the most significant public health advancement in human history.

How chlorination works:

**Potent Germicide Reduction** in the level of many disease-causing microorganisms in drinking water to almost immeasurable levels.

**Taste and Odor Reduction** of many disagreeable tastes and odors like foul-smelling algae secretions, sulfides, and odors from decaying vegetation.

**Biological Growth Elimination** of slime bacteria, molds, and algae that commonly grow in water supply reservoirs, on the walls of water mains, and in storage tanks.

**Chemical Removal** of hydrogen sulfide (which has a rotten egg odor), ammonia, and other nitrogenous compounds that have unpleasant tastes and hinder disinfection. It also helps to remove iron and manganese from raw water.



## Sampling Results

During the past year we have taken hundreds of water samples in order to determine the presence of any radioactive, biological, inorganic, volatile organic or synthetic organic contaminants. The table below shows only those contaminants that were detected in the water. The state requires us to monitor for certain substances less than once per year because the concentrations of these substances do not change frequently. In these cases, the most recent sample data are included, along with the year in which the sample was taken.

We participated in the 3rd stage of the EPA’s Unregulated Contaminant Monitoring Regulation (UCMR3) program by performing additional tests on our drinking water. UCMR3 benefits the environment and public health by providing the EPA with data on the occurrence of contaminants suspected to be in drinking water, in order to determine if EPA needs to introduce new regulatory standards to improve drinking water quality.

REGULATED SUBSTANCES							
SUBSTANCE (UNIT OF MEASURE)	YEAR SAMPLED	MCL [MRDL]	PHG (MCLG) [MRDLG]	AMOUNT DETECTED	RANGE LOW-HIGH	VIOLATION	TYPICAL SOURCE
Arsenic <sup>1</sup> (ppb)	2013	10	0.004	6	2–9	No	Erosion of natural deposits; runoff from orchards; glass and electronics production wastes
Barium (ppb)	2012	1,000	2,000	52	23–84	No	Discharges of oil drilling wastes and from metal refineries; erosion of natural deposits
Chlorine (ppm)	2013	[4.0 (as Cl <sub>2</sub> )]	[4 (as Cl <sub>2</sub> )]	0.8	0.3–2.0	No	Drinking water disinfectant added for treatment
Fluoride (ppm)	2012	2.0	1	<1	<1–<1	No	Erosion of natural deposits; water additive that promotes strong teeth; discharge from fertilizer and aluminum factories
Gross Alpha Particle Activity (pCi/L)	2007	15	(0)	<3	<3–5	No	Erosion of natural deposits
Haloacetic Acids–Stage 2 (ppb)	2013	60	NA	3	<1–7	No	By-product of drinking water disinfection
Heterotrophic Plate Count Bacteria (CFU/ml)	2012	Surface water treatment = TT	HPC = NA; Others = (0)	41	<1–140	No	Naturally present in the environment
Nitrate [as nitrate] <sup>2</sup> (ppm)	2013	45	45	29	8–40	No	Runoff and leaching from fertilizer use; leaching from septic tanks and sewage; erosion of natural deposits
Selenium (ppb)	2012	50	30	<2	<2–3	No	Discharge from petroleum, glass, and metal refineries; erosion of natural deposits; discharge from mines and chemical manufacturers; runoff from livestock lots (feed additive)
TTHMs [Total Trihalomethanes]–Stage 2 (ppb)	2013	80	NA	26	3–60	No	By-product of drinking water disinfection
Turbidity (NTU)	2012	TT	NA	3	<1–3	No	Soil runoff
Tap water samples were collected for lead and copper analyses from sample sites throughout the community							
SUBSTANCE (UNIT OF MEASURE)	YEAR SAMPLED	AL	PHG (MCLG)	AMOUNT DETECTED (90TH%TILE)	SITES ABOVE AL/ TOTAL SITES	VIOLATION	TYPICAL SOURCE
Copper (ppb)	2012	1,300	300	16	0/42	No	Internal corrosion of household plumbing systems; erosion of natural deposits; leaching from wood preservatives
Lead (ppb)	2012	15	0.2	ND	0/42	No	Internal corrosion of household water plumbing systems; discharges from industrial manufacturers; erosion of natural deposits

SECONDARY SUBSTANCES							
SUBSTANCE (UNIT OF MEASURE)	YEAR SAMPLED	SMCL	PHG (MCLG)	AMOUNT DETECTED	RANGE LOW-HIGH	EXCEEDANCE	TYPICAL SOURCE
Color (Units)	2012	15	NS	1	1–5	No	Naturally-occurring organic materials
Iron <sup>3</sup> (ppb)	2012	300	NS	<50	<50–450	No	Leaching from natural deposits; industrial wastes
Manganese <sup>3</sup> (ppb)	2012	50	NS	<10	<10–160	No	Leaching from natural deposits
Total Dissolved Solids (ppm)	2012	1,000	NS	414	290–720	No	Runoff/leaching from natural deposits

## UNREGULATED AND OTHER SUBSTANCES

SUBSTANCE (UNIT OF MEASURE)	YEAR SAMPLED	AMOUNT DETECTED	RANGE LOW-HIGH	TYPICAL SOURCE
<b>1,2,3-Trichloropropane</b> <sup>4</sup> (ppt)	2013	108	70–320	TCP detections in the Central Valley of California are being attributed to past use of soil fumigants to battle nematodes
<b>Hardness</b> (ppm)	2012	137	83–260	Erosion of natural deposits
<b>pH</b> (Units)	2012	8.2	7.2–8.8	Inherent characteristics of water

## Definitions

**AL (Regulatory Action Level):** The concentration of a contaminant which, if exceeded, triggers treatment or other requirements that a water system must follow.

**MCL (Maximum Contaminant Level):** The highest level of a contaminant that is allowed in drinking water. Primary MCLs are set as close to the PHGs (or MCLGs) as is economically and technologically feasible. Secondary MCLs (SMCLs) are set to protect the odor, taste and appearance of drinking water.

**MCLG (Maximum Contaminant Level Goal):** The level of a contaminant in drinking water below which there is no known or expected risk to health. MCLGs are set by the U.S. EPA.

**MRDL (Maximum Residual Disinfectant Level):** The highest level of a disinfectant allowed in drinking water. There is convincing evidence that addition of a disinfectant is necessary for control of microbial contaminants.

**MRDLG (Maximum Residual Disinfectant Level Goal):** The level of a drinking water disinfectant below which there is no known or expected risk to health. MRDLGs do not reflect the benefits of the use of disinfectants to control microbial contaminants.

**NA:** Not applicable

**ND (Not detected):** Indicates that the substance was not found by laboratory analysis.

**NS:** No standard

**NTU (Nephelometric Turbidity Units):** Measurement of the clarity, or turbidity, of water. Turbidity in excess of 5 NTU is just noticeable to the average person.

**pCi/L (picocuries per liter):** A measure of radioactivity.

**PDWS (Primary Drinking Water Standard):** MCLs and MRDLs for contaminants that affect health along with their monitoring and reporting requirements, and water treatment requirements.

**PHG (Public Health Goal):** The level of a contaminant in drinking water below which there is no known or expected risk to health. PHGs are set by the California EPA.

**ppb (parts per billion):** One part substance per billion parts water (or micrograms per liter).

**ppm (parts per million):** One part substance per million parts water (or milligrams per liter).

**ppt (parts per trillion):** One part substance per trillion parts water (or nanograms per liter).

**TT (Treatment Technique):** A required process intended to reduce the level of a contaminant in drinking water.

<sup>1</sup> While your drinking water meets the federal and state standard for arsenic, it does contain low levels of arsenic. The arsenic standard balances the current understanding of arsenic's possible health effects against the costs of removing arsenic from drinking water. The U.S. EPA continues to research the health effects of low levels of arsenic, which is a mineral known to cause cancer in humans at high concentrations and is linked to other health effects such as skin damage and circulatory problems.

<sup>2</sup> Nitrate in drinking water at levels above 45 ppm is a health risk for infants of less than six months of age. Such nitrate levels in drinking water can interfere with the capacity of the infant's blood to carry oxygen, resulting in a serious illness; symptoms include shortness of breath and blueness of the skin. Nitrate levels above 45 ppm may also affect the ability of the blood to carry oxygen in other individuals, such as pregnant women and those with certain specific enzyme deficiencies. If you are caring for an infant, or you are pregnant, you should ask advice from your health care provider.

<sup>3</sup> In May 2012, a testing result of 450 ppb for iron and 160 ppb for manganese occurred at one well after a cleaning and maintenance project. The elevated levels are being attributed to the well not being completely flushed. Secondary MCLs are established to regulate the aesthetics of drinking water like taste and odor.

<sup>4</sup> In 1999, the state established a TCP Notification Level (NL) for water systems of 5 ppt. If a water source has detections of more than 100 times the NL, which calculates to 500 ppt, then the state recommends that the source be removed from service. Our TCP levels have never exceeded 500 ppt. Since July 2004, the state has been working to establish an MCL for TCP, which is now expected to be released for public comment this year. To respond to the coming MCL, the Shafter City Council directed its Public Works Department to be proactive and work toward having treatment facilities added to our well sites to remove TCP from our water. The state's Public Health Goal (PHG) for TCP is 0.7 ppt. Some people who use water containing TCP in excess of the PHG and NL over many years may have an increased risk of getting cancer, based on studies on laboratory animals.